# metal-organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 295 KMean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$ Disorder in main residue R factor = 0.025 wR factor = 0.076 Data-to-parameter ratio = 16.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# *trans*-Diaqua(malonato)(1,10-phenanthroline)nickel(II) trihydrate

The title complex,  $[Ni(C_3H_2O_4)(C_{12}H_8N_2)(H_2O)_2]\cdot 3H_2O$ , displays a distorted octahedral NiN<sub>2</sub>O<sub>4</sub> coordination geometry, formed by a malonate dianion, a 1,10-phenanthroline ligand and two *trans* water molecules. The Ni atom and one O atom, the latter from a solvent water molecule, are located on twofold axes of symmetry. The crystal structure is stabilized by extensive  $O-H\cdots O$  hydrogen-bonding interactions. Received 21 October 2005 Accepted 1 November 2005 Online 10 November 2005

## Comment

The malonate anion (mal) displays a variety of coordination modes owing to the presence of two carboxylate groups in the 1- and 3-positions (Catalina *et al.*, 2000*a,b*; Yolanda, Catalina *et al.*, 2001; Yolanda, Joaquín *et al.*, 2001; Yilmaz *et al.*, 2004). The analogous *cis*-Ni<sup>II</sup> complex containing malonate and 1,10-phenanthroline (phen) has been reported (Wang *et al.*, 2004). In this paper, the title Ni<sup>II</sup> complex, (I), with a *trans* configuration, has been prepared and its crystal structure determined.



The crystal structure of (I) contains the [Ni(mal)(phen)- $(H_2O)$ ] complex and uncoordinated water molecules, in a 1:3 ratio, as shown in Fig. 1. The Ni atom is located on a twofold axis, as is one of the water molecules, *viz*. that containing the O5 atom. The Ni<sup>II</sup> centre is six-coordinated by two carboxylate O atoms from the malonate dianion, two N atoms of the phen ligand, and two O atoms of the coordinated water molecules. Both the dianion and neutral ligands chelate the Ni centre; the water molecules occupy *trans* positions in the distorted octahedral geometry. The Ni–O and Ni–N bond lengths (Table 1) are almost equal to those of the aforementioned *cis*-Ni<sup>II</sup> complex (Wang *et al.*, 2004).

The crystal structure of (I) appears to be tightly consolidated by extensive hydrogen bonds, as detailed in Table 2. Neighbouring complexes are connected to each other *via* O–  $H \cdots O$  hydrogen bonds. Atom H5 is involved in a bifurcated O– $H \cdots (O,O)$  hydrogen bond; the sum of the angles about this atom is 359°.

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# Experimental

Nickel chloride hexahydrate (0.0475 g, 0.2 mmol) and malonic acid (0.0208 g, 0.2 mmol) were dissolved in water (20 ml). The resulting solution was added to an absolute ethanol solution (10 ml) containing 1,10-phenanthroline (0.0396 g, 0.2 mmol), and stirred for 5 h until a small amount of precipitate appeared. This solution was filtered and the filtrate was set aside at room temperature. After three weeks, pale-blue single crystals were obtained.

 $D_x = 1.558 \text{ Mg m}^{-3}$ 

Cell parameters from 4010 reflections  $\theta = 2.5-27.4^{\circ}$  $\mu = 1.11 \text{ mm}^{-1}$ T = 295 (1) K Block, pale blue  $0.40 \times 0.37 \times 0.30 \text{ mm}$ 

2097 independent reflections

 $R_{\rm int}=0.016$ 

 $\begin{array}{l} \theta_{\max} = 27.4^{\circ} \\ h = -11 \rightarrow 11 \end{array}$ 

 $k = -15 \rightarrow 15$ 

 $l = -12 \rightarrow 12$ 

1893 reflections with  $I > 2\sigma(I)$ 

Mo  $K\alpha$  radiation

### Crystal data

$[Ni(C_3H_2O_4)(C_{12}H_8N_2)-$
$(H_2O)_2]\cdot 3H_2O$
$M_r = 431.04$
Monoclinic, $P2/n$
a = 8.9389 (3) Å
b = 12.1466 (4) Å
c = 9.4334 (3) Å
$\beta = 116.189 \ (1)^{\circ}$
V = 919.11 (5) Å <sup>3</sup>
Z = 2

#### Data collection

Rigaku R-AXIS RAPID diffractometer  $\omega$  scans Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)  $T_{min} = 0.632, T_{max} = 0.717$ 4010 measured reflections

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0426P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.025$	+ 0.2459P]
$wR(F^2) = 0.076$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\rm max} = 0.001$
2097 reflections	$\Delta \rho_{\rm max} = 0.34 \text{ e} \text{ Å}^{-3}$
129 parameters	$\Delta \rho_{\rm min} = -0.38 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
	Extinction coefficient: 0.054 (3)

#### Table 1

Selected geometric parameters (Å, °).

Ni-O1 Ni-O3	2.0042 (12) 2.1111 (12)	Ni-N1	2.0714 (13)
O1-Ni-O3	90.58 (5)	O3-Ni-O3 <sup>i</sup>	177.11 (6)
O1-Ni-N1	93.65 (5)	O3-Ni-N1 <sup>1</sup>	90.69 (5)
O1-Ni-O1 <sup>1</sup>	92.35 (7)	N1-Ni-O3	87.10 (5)
O1-Ni-N1 <sup>i</sup>	173.83 (5)	N1-Ni-N1 <sup>i</sup>	80.38 (7)
$O1-Ni-O3^{i}$	91.42 (5)		

Symmetry code: (i)  $-x + \frac{3}{2}, y, -z + \frac{3}{2}$ .

## Table 2

#### Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O3-H31O2 <sup>ii</sup>	0.81	1.98	2.7580 (18)	162
O3−H32···O4	0.91	1.85	2.748 (2)	171
O4-H41···O2 <sup>iii</sup>	0.93	1.97	2.897 (2)	177
O4−H42···O5	0.95	1.88	2.824 (2)	175
$O5-H5\cdots O2^{iv}$	0.88	1.97	2.846 (2)	173
$O5-H5\cdots O1^{iv}$	0.88	2.54	3.1637 (15)	129

Symmetry codes: (ii)  $x + \frac{1}{2}, -y, z + \frac{1}{2}$ ; (iii) -x + 1, -y, -z + 2; (iv)  $-x + \frac{1}{2}, y, -z + \frac{3}{2}$ .



### Figure 1

The molecular structure of (I), shown with 40% probability displacement ellipsoids. Both sites of the disordered C8 atom are shown. [Symmetry code: (i) 1.5 - x, y, 1.5 - z.]

Aromatic H atoms were placed in calculated positions, with C–H = 0.93 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$ . The H atoms of the water molecules were located in difference Fourier maps and and made to ride on the O atoms in their as-found relative positions,  $U_{iso}(H) = 0.05 \text{ Å}^2$ . Atom C8 of the malonate ligand was found to be disordered over two positions about the twofold axis. This atom was refined anisotropically, with 0.50 site-occupancy factors, and the H atoms were were idealized and made to ride with C–H = 0.97 Å and  $U_{iso}(H) = 1.2U_{eq}(C8)$ .

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/ MSC, 2002); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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